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And

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Chapter2 Acids Bases And Salts

Acid + base \rightarrow salt +
water + heat. H_2SO_4
 $+MgO \rightarrow MgSO_4 +H_2O$
 $2HCl+Mg(OH)_2$
 $\rightarrow MgCl_2 +2H_2O$. 2.

Reaction of non-metal
oxides with bases. Non-
metal oxides are acidic
in nature Base + Non-
metal oxide \rightarrow salt +
water + heat.

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$\text{CO}_3 + \text{H}_2\text{O}$. To know more about Properties of Acids and Bases, visit here. Water Acids and bases in water

Class 10 Chapter 2
Science Notes Acids,
Bases, and Salts

The reaction between an acid and a base to give a salt and water is known

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as a neutralisation reaction. In general, a neutralisation reaction can be written as $\text{Base} + \text{Acid} \rightarrow \text{Salt} + \text{Water}$

2.1.5 Reaction of Metallic Oxides with Acids

Activity 2.7 n

Take a small amount of copper oxide in a beaker and add dilute hydrochloric acid slowly while stirring.

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CHAPTER2 Acids, Bases and Salts

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Notes Chapter 2 Acids
Bases and Salts. 1.

Common Salt (Sodium
Chloride): Sodium
chloride (NaCl) is also
known as Common or
Table Salt. It is formed
after the reaction
between sodium ... 2.

Bleaching Powder

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(CaOCl_2): Bleaching powder is also known as chloride of lime. It is a ...

Acids Bases and Salts
Class 10 Notes Science
Chapter 2 ...

Notes Chapter - 2 Acids,
Bases and Salts ACIDS:

□ These are the substances which have sour in taste. □ They

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Chapter2 Acids

turns blue litmus

solution to red. □ They give H ions in aqueous solution Examples of

Acids : - HCl -

Hydrochloric Acid

BASES □ These

substances are bitter in taste. □ They turns red litmus solution to blue.

Chapter - 2, Acids Bases
and Salts Notes

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Chapter 2 Notes Acids

Bases and Salts Acids

Bases and Salts Class 10

Notes Understanding

the Lesson 1. Acids and

Bases: Acids are sour in

taste and change the

colour of blue litmus to

red.

Acids Bases and Salts

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Chapter 2 And

Salts. When acid and base neutralize, salts are formed. Strong acid and strong base combines to form neutral salt. $\text{NaOH} + \text{HCl} \rightarrow \text{NaCl} + \text{H}_2\text{O}$.

Eq.1. Formation of Neutral Salt. Strong acid and weak base combine to form acidic salt. For Example, Hydrochloric Acid and ammonium hydroxide combine to

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form ammonium chloride. Other examples, sodium hydrogen carbonate, sodium hydrogen sulphate etc.

Revision Notes for
Science Chapter 2 -
Acids, bases and ...
Study Material and
Notes of Ch 2 Acids,
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10th Science. The sour and bitter tastes of food are due to acids and bases are present in them. Acids are sour in taste and change the colour of blue litmus to red. Litmus solution is a purple dye, which is extracted from lichen.

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10th Science And

Chemicals from
common salt; NCERT
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Chapter 2 Science Acids
bases and salts. In this
chapter, students get
hold of basic knowledge
on Acids bases and
salts. In this chapter,
various chemical
properties of acids,
bases and salts and the
reaction of them with

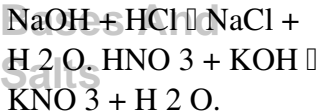
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metals, non-metals are
studied.

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Acids, bases and salt
When an acid and base
react with each other to
give a salt and water, it
is termed as
neutralisation reaction
i.e., Acid + Base \rightarrow Salt +
Water For example:

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Chapter 2 Acids



Question 15. Give two important uses of washing soda and baking soda. Answer: Important uses of washing soda are:

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Science Chapter 2 Notes
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been provided by
India's topmost
Chemistry teachers.
Also in this article, you
will find the Step-wise
explanation for each and
every question.

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Chapter 2 Acids ...

(ii) Acids react with metal carbonate and metals hydrogen carbonate to give salt, water and carbon dioxide. (iii) Acids react with bases to give salt and water. This reaction is called neutralization

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Bases And Acids
react with metals oxides
to give salt and water.

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Questions with Answers
Acids, Bases and Salts
Acids, Bases and Salts
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Questions Very Short
Answer Type Question

1. How is the concentration of hydronium (H_3O^+) ions affected when a solution of an acid is diluted? (2017 OD)

Acids, Bases and Salts
Class 10 Important
Questions with ...

Video Timeline: 0:00:00

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start 0:00:21 Difference
between Acids & bases

0:10:38 what are
indicators? 0:17:07

Universal Indicators

0:21:11 Salts 0:24:51 a)

Sod...

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Heya!!! TAta NAMAK

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Salts dissociate in water
fully and form their
basic and acidic

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counterpart. For

example here

CH_3COONa will form CH_3COOH and NaOH with water. Since NaOH is highly basic, the solution will be basic.

N.b. the reverse of this also applies. For example, a solution of a salt of a strong acid and weak base will be acidic.

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Questions Very Short
Answer Type. Question

1. Two solutions A and B have pH values of 5 and 8 respectively.

Which solution will be basic in nature ...

Acids Bases and Salts

Class 10 Extra

Questions with ...

For example,

Hydrochloric acid

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Chapter 2 Acids

(HCl), Sulphuric acid (H_2SO_4), Nitric acid (HNO_3). Bases - Bases are compounds which give hydroxide ion in water solution. For example, Sodium hydroxide (NaOH), Potassium hydroxide (KOH), Calcium hydroxide ($Ca(OH)_2$). Salts - Salts are compounds obtained by replacement of H from

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