

## Chemquest 38 Answer Key

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$\text{Cu}(\text{NO}_3)_2 + \text{Zn} \rightarrow \text{Zn}(\text{NO}_3)_2 + \text{Cu}$  There is a sentence that you can write to describe what happens above: Copper(II) nitrate reacts with zinc metal and yields zinc nitrate and copper metal. Take a minute to notice the relationship between the sentence and the symbols in the reaction

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ChemQuest 23 Name: \_\_\_\_\_ Date: \_\_\_\_\_ Hour: \_\_\_\_\_ Information: Drawing ... Hint: add your answer to question 1 and four times your answer to question 2.  $4 + 4(7) = 4 + 28 = 32$  4. Verify that there are 32 electrons pictured in the drawing of  $\text{CCl}_4$ . Count each dot in the picture above: there are 24. There are also 8 more electrons represented by the 4 lines (each line represents two electrons). 5 ...

ChemQuest 23 : Drawing Covalent Compounds

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