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An Introduction to Thermal Physics

1. DIESEL ENGINES 2. As in the Otto cycle,
the red and blue curves are adiabats, so no
heat is exchanged during these steps. The
efficiency of the Diesel cycle is most easily
worked out by calculating the heat input Q
along the yellow step 2 to 3 and the heat
expelled during the green step 4 to 1. The
efficiency is then $e = W/Q$.

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DIESEL ENGINES An Introduction to Thermal Physics

in Schroeder), $\Delta H_{H_2O} = 285.83 \text{ kJ mol}^{-1}$ and $\Delta H_{H_2} = 890.36 \text{ kJ mol}^{-1}$ (12) This time, the final volume is $\frac{1}{3}$ of the initial volume, since the 2 moles of water has condensed out as liquid with negligible volume compared to the gases. Thus the atmosphere does work $P\Delta V = RT\Delta n$ (13) $= 8.31 \text{ J K}^{-1} (298 \text{ K}) (2)$ (14) $= 4.953 \text{ kJ}$ (15)

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