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Laboratory manual for physical chemistry where T_0 is the freezing point of the solvent. If BC in Fig. 1st-1 is taken to be a straight line, the slope dp_A/dT may be replaced by CB/BD , i.e., by $(p_A - p_s)/\Delta T$, so that $p_p T H_p R T A s m v_A 0 2 \Delta = \Delta \Delta 0$ Eq. 1st.4 The slope of the vapor pressure curve of the solid solvent is also given by another form of the

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Laboratory manual for physical chemistry where T_0 is the freezing point of the solvent If BC in Fig 1st-1 is taken to be a straight line, the slope dp_A/dT may be replaced by CB/BD , ie, by $(p_A - p_s)/\Delta T$, so that $p_p T H_p R T A s m v_A 0 2 \Delta = \Delta \Delta 0$ Eq 1st4 The slope

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June 2018; ISBN: 978-93-5311-329-2;
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the temperature reaches a steady reading.
Record the temperature and the
equilibrium pressure. Increase the pressure
in the system by 3-5 mmHg (around 6
kPa) by allowing a small amount of air to
enter through tap B slowly.

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Chem 343 Physical Chemistry Laboratory Manual Spring 2014 – ADH 1/10/14 By Audrey Dell Hammerich, Luke Hanley, and Preston Snee with contributions from Yoshitaka Ishii, Sanja Tepavcevic, and Ben Stokes 1

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